

book for completeness, and by this I mean not thoroughness in breadth and depth of coverage, but how completely has the author lived up to the objectives set forth in his preface. (A major objective of the books in this series is to have each serve as a *vade mecum* to a particular field of chemistry.)

This particular volume is, of course, two books; one on the Alkali Metals and the other on Hydrogen. I was impressed and quite pleased with the section on the Alkali Metals, written by Prof. John F. Suttle. I found no areas of disagreement on those subjects I was familiar with, and I feel I have been introduced graciously and well to unfamiliar subjects by Prof. Suttle's lucid exposition and easy-to-read style of writing. Of particular importance to any technical guide are its references to the pertinent technical literature, both survey and specific, and I consider this section to be particularly well documented. Coverage of the general field of the chemistry of the alkali metals is certainly extensive and, I believe, well balanced in depth. I feel the author has done an outstanding job in achieving the objectives set forth by the editors of this series.

I was somewhat disappointed in the section on Hydrogen, written by Prof. Brasted. Not that it is bad, for it isn't, but I feel it could have been considerably better. Most of the information presented appears to be valid data, with the exception of a few instances in which out-dated and erroneous material from the literature has been used, but the emphasis on relatively trivial points and the omission or relative neglect of important subjects is disturbing. For example, this section is quite short, comprising less than 20% of the book, yet ten pages out of the forty are devoted to methods for the preparation of hydrogen and, in spite of the preface telling us that this book is to be a guide for advanced workers, much of this discussion of preparative methods is very elementary and of the sort found in introductory high school chemistry books. On the other hand, there is no discussion in this section of the chemical behavior of hydrogen ion in solution, *i.e.*, acids and bases, and the discussion of the chemical bonding of hydrogen is exceedingly skimpy. Other examples could be cited of elaboration of trivia and neglect of the significant. I also noted the absence of certain definitive or survey references to the literature which would be of considerable help to those seeking further information on particular subjects. For example, in the short discussion of hydrogen in metals, there is no reference to Smith's book on "Hydrogen in Metals."

To sum up: the section on the Alkali Metals more than makes up for the shortcomings of the section on Hydrogen, and the book will be a valuable addition to any chemist's bookshelf. I'm happy to have it.

GENERAL ELECTRIC CO.
CLEVELAND, OHIO

DALLAS T. HURD

Médicaments de Synthèse. By H. P. KAUFMANN, Directeur de l'Institut de Chimie pharmaceutique et de Technologie chimique de l'Université de Munster. Masson et Cie., Editeurs, 120, Boulevard Saint-Germain, Paris VIe, France. 1957. viii + 853 pp. 17 × 25.5 cm. Price, 11.500 Fr.

The organization of this monograph is based essentially on the physiological functions of the human organ systems. Classes and groups of chemical agents are arranged in accordance with their influence on such functions. Thus the principal chapters comprise: substances acting on the central nervous system, as narcotics, hypnotics and spasmolytics; substances acting on the peripheral nervous system, as local anaesthetics; substances acting on the cardiovascular system; diuretics; substances acting on the gastrointestinal tract; antimittic agents; vitamins and hormones; disinfectants and chemotherapeutic (in the classic sense) agents; and antibiotics.

Each chapter is introduced with a rather extensive discussion of anatomical, physiological and pharmacological aspects of a given activity; illustrations with pictures, drawings and schematic sketches will be greatly appreciated by readers with no formal medical education.

The book is obviously and primarily directed to the pharmacist and pharmaceutical chemist. Preparative and synthetic aspects of man-made and natural products form the core of this treatise, as indicated by its title. Practically all the chapters are concluded with well arranged tabulations of structural formulas, series of closely related deriva-

tives, trivial and trade names, clinical indications and dosages of medication. Efforts have been made to approach and elucidate the mechanism of action, and to correlate structural and functional features of the chemical agents with their specific activities. In this connection it must be regarded unfortunate that so little attention has been paid to the modern theories of antimetabolite action. It also is felt that more discretion might have been exercised in deciding between important and unimportant, and modern and antiquated. Many readers would probably have preferred seeing the chapters on antipyretics, antimalarials, sympathomimetic agents, the vitamins (particularly the B complex) with their long preparative procedures shortened in favor of, *e.g.*, the cardioactive glucosides, the tranquilizing drugs and the so-called cancer-chemotherapeutic agents which have been allotted unduly little or no space at all. The somewhat unfortunate policy of omitting the double bonds in the benzene nucleus will be misleading in a number of instances as with Prominal and Evipan (p. 47) or with Ritalin and the diethylaminoethylamido analog of Dolantin (p. 105) where the phenyl group is depicted as a benzene ring and a cyclohexane ring, respectively. Confusing are the formulas of Cignolin (p. 635) of Parpanit (p. 123), Spasmodex and Artane (p. 124). Furthermore, a moderate number of mistakes and misprints have been overlooked: see eupaverine (p. 116), the antique formula of santonin (p. 272), the formulas of lumisterol and pyrocalciferol (p. 392), of androsterone (p. 419), of digitoxigenin (p. 485), of pseudodiosgenin diacetate (p. 486) and the steroidal formulas of the tabulations (pp. 493-496) where the indication of the allo configuration is missing.

"Médicaments de Synthèse" is a translation of H. P. Kaufmann's "Arzneimittel Synthese (Springer Verlag 1953)" by the well known French chemist F. Wintermütz who has succeeded in bringing this monograph completely up to date by including the very recent literature as, *e.g.*, pertaining to the morphine alkaloids, the antibiotics and, in particular, the steroidal hormones. This is no small achievement in a field of such vastness and rapid development. The monograph is furnished with an author index, a most detailed and complete subject index and an up-to-date index of American and European patents. It will bring a wealth of information to students and teachers of pharmaceutical and medicinal chemistry, to men engaged in academic and particularly industrial pharmaceutical research and finally to all the practitioners and clinicians who have more than a casual interest in the chemistry and pharmacology of the drugs at their disposal.

NATIONAL INSTITUTES OF HEALTH, NIAMD
BETHESDA 14, MARYLAND

ERICH MOSETTIG

Some Principles of Energetics in Biochemical Reactions.

By IRVING M. KLOTZ, Professor of Chemistry, Northwestern University, Evanston, Illinois. Academic Press Inc., 111 Fifth Avenue, New York 3, N. Y. 1957. vii + 64 pp. 14.5 × 22 cm. Price, \$3.00.

This book is intended for biologists without knowledge of even elementary concepts of physical chemistry. It introduces them, in the space of 64 pages, to a variety of topics: the first and second laws of thermodynamics, statistical mechanics, free energy and its dependence on concentration, the conversion of ATP to ADP, electrochemistry, osmotic pressure, sedimentation equilibrium, the melting of ice, the stretching of rubber, protein denaturation, etc. The object is to give the biologist a "reading knowledge" of thermodynamics with which he is supposed to be able to "understand the acknowledged classics in the field."

The title of the book suggests that biochemical reactions may be given a somewhat detailed treatment, but this is not the case. For instance, in the discussion of the conversion of ATP to ADP, there is no mention of how one computes the activity of the solvent, though water appears in the reaction equation. Nor is it pointed out that pH may alter the nature of the participating ions.

Presumably there must be biologists who will derive benefit from this book's cursory glance at thermodynamics. Biochemists, however, will require a much deeper treatment, such as is contained in Professor Klotz's "Chemical Thermodynamics" or a text of equivalent stature.

DEPARTMENT OF CHEMISTRY
STATE UNIVERSITY OF IOWA
IOWA CITY, IOWA

CHARLES TANFORD